

FACULTY OF ENGINEERING
Scheme of Instruction & Examination
(AICTE Model Curriculum for the Academic Year 2018-2019)
and
Syllabi
B.E. I and II Semesters (Group-B)
of
Four Year Degree Programme
in
B.E. (Common to All Branches)
(With effect from the Academic Year 2018– 2019)
(As approved in the Faculty Meeting held on 26th June 2018)



Issued by
Dean, Faculty of Engineering
Osmania University, Hyderabad
2018

GROUP DISTRIBUTION

B.E. (I, II – Semesters)

**NUMBER OF DIVISIONS PER COURSE OF
O.U. AFFILIATED RESPECTIVE ENGINEERING COLLEGES**

S. No	COLLEGE NAME	GROUP – A					No. of Div.	GROUP – B				No. of Div.	Total No. of Div.
		ECE	IT	ME	PE	AE		CSE	CE	EEE	EIE		
1	MVSR	3	2	2	-	1	8	3	2	2	-	7	15
2	MJCET	2	2	2	1	-	7	2	2	1	1	6	13
3	DCET	2	1	2	1	-	6	2	2	1	1	6	12
4	ISL	2	1	1	-	-	4	2	2	1	-	5	9
5	METHODIST	2	-	2	-	-	4	2	2	1	-	5	9
6	MEC	2	-	1	-	-	3	2	1	1	-	4	7
7	SWATHI	1	-	1	-	-	2	1	1	-	-	2	4
8	STANLEY	2	1	-	-	-	3	3	-	1	-	4	7
9	NGIT	-	2	-	-	-	2	3	-	-	-	3	5
10*	NSAKCET	2	1	4	-	-	7	2	3	1	-	6	13
11*	LORDS	1	1	4	-	-	6	2	3	1	-	6	12
	TOTAL	19	11	19	2	1	52	24	18	10	2	54	106

Note: * Applied to OU for Affiliation from the academic year 2019-2020

Group – B

CSE : Computer Science and Engineering

CE : Civil Engineering

EEE : Electrical & Electronics Engineering

EIE : Electronics & Instrumentation Engineering

SCHEME OF INSTRUCTION & EXAMINATION
B.E. (All Branches) I - Semester
(Group B – CSE, CE, EEE, EIE)

S. No.	Course Code	Course Title	Scheme of Instructions				Scheme of Examination			Credits
			L	T	P/ D	Contact Hours/Week	CIE	SEE	Duration in Hours	
MC : Three Week Induction Programme										
Theory Course										
1	MC112CE	Environmental Science	2	-	-	2	30	70	3	-
2	MC113PY	Essence of Indian Traditional Knowledge	2	-	-	2	30	70	3	-
3	BS102MT	Mathematics-I	3	1	-	4	30	70	3	4
4	BS105CH	Chemistry	3	1	-	4	30	70	3	4
5	ES107CS	Programming for Problem Solving	3	-	-	3	30	70	3	3
Practical/ Laboratory Course										
6	BS153CH	Chemistry Lab	-	-	3	3	25	50	3	1.5
7	ES155CS	Programming for Problem Solving Lab	-	-	4	4	25	50	3	2
8	ES157ME	Workshop/ Manufacturing Process	1	-	4	5	50	50	3	3
Total			14	02	11	27	250	500		17.5

BS: Basic Science**ES:** Engineering Science**MC:** Mandatory Course**L:** Lecture**T:** Tutorial**P:** Practical**D:** Drawing**CIE:** Continuous Internal Evaluation**SEE:** Semester End Examination (Univ. Exam)**PY:** Philosophy**MT:** Mathematics**CH:** Chemistry**CE:** Civil Engineering, **CS:** Computer Science and Engineering, **ME:** Mechanical Engineering.**Note:**

- Each contact hour is a Clock Hour.
- The duration of the practical class is two hours, however it can be extended wherever necessary, to enable the student to complete the experiment.

Course Code	Course Title					Core/Elective	
MC112CE	Environmental Science (Common to All Branches)					Mandatory Course	
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	2	-	-	-	30	70	-
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ To create awareness and impart basic knowledge about the environment and its allied problems. ➤ To know the functions of ecosystems. ➤ To understand importance of biological diversity. ➤ To study different pollutions and their impact on environment. ➤ To know social and environment related issues and their preventive measures. <p>Course Outcomes</p> <p>After completing this course, the student will be able to:</p> <ol style="list-style-type: none"> 1. Adopt environmental ethics to attain sustainable development. 2. Develop an attitude of concern for the environment. 3. Conservation of natural resources and biological diversity. 4. Creating awareness of Green technologies for nation's security. 5. Imparts awareness for environmental laws and regulations. 							

UNIT-I

The Multidisciplinary Nature of Environmental Studies: Definition, scope and importance, need for public awareness.

Natural Resources: Water Resources – Use and over utilization of surface and ground water, flood, drought, conflicts over water, Dams: Benefits and Problems. Food Resources –World Food Problems, effects of modern agriculture, fertilizer-pesticides problems, water logging, salinity, Forest Resources – Use and over exploitation, deforestation & its effect on tribal people. Land Resources –Land Degradation, environmental effect of mining, man induced landslides, soil erosion and desertification. Energy Resources –Growing energy needs, Renewable and Non-renewable energy resources.

UNIT-II

Ecosystems: Concept of an ecosystem, Structure and function of an ecosystem, Producers, consumers and decomposers, Energy flow in ecosystem, food chains, ecological pyramids, ecological succession, types of ecosystems (marine, pond, river, forest, grassland, desert)

UNIT-III

Biodiversity: Levels of Biodiversity, Bio-geographical classification of India, Value of biodiversity, Threats to biodiversity, endangered and endemic species of India, Conservation of biodiversity, global and national efforts.

UNIT-IV

Environmental Pollution: Definition, Causes, effects and control measures of air pollution, water pollution, soil pollution, noise pollution, thermal pollution, solid waste management.

Environment Protection Act: Air, water, forest and wildlife Acts, issues in the enforcement of environmental legislation.

UNIT-V

Social Issues and the Environment: Watershed management and environmental ethics. Climate change, global warming, acid rain, ozone layer depletion.

Environmental Disaster Management: Types of disasters, impact of disasters on environment, infrastructure, and development. Basic principles of disaster mitigation, disaster management, and methodology. Disaster management cycle and disaster management in India.

Field Work:

- Visit to a local area to document environmental issues- agricultural area/ pond/lake/terrestrial ecosystem
- Visit to a local polluted area- market/slum area/Industrial area/traffic area

Suggested Reading:

1. A.K. De, *Environmental Chemistry*, Wiley Eastern Ltd.
2. E.P. Odum, *Fundamentals of Ecology*, W.B. Saunders Co., USA.
3. M.N. Rao and A.K. Datta, *Waste Water Treatment*, Oxford and IBK Publications.
4. Benny Joseph, *Environmental Studies*, Tata McGraw Hill, 2005.
5. V.K. Sharma, *Disaster Management*, National Centre for Disaster Management, IIPe, 1999.

Course Code	Course Title					Core/Elective	
MC113PY	Essence of Indian Traditional Knowledge (Common to All Branches)					Mandatory Course	
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	2	-	-	-	30	70	-
<p>Course Objectives The course will introduce the students to</p> <ul style="list-style-type: none"> ➤ To get a knowledge in Indian Culture ➤ To Know Indian Languages and Literature and the fine arts in India ➤ To explore the Science and Scientists of Medieval and Modern India <p>Course Outcomes After successful completion of the course the students will be able to</p> <ol style="list-style-type: none"> 1. Understand philosophy of Indian culture. 2. Distinguish the Indian languages and literature. 3. Learn the philosophy of ancient, medieval and modern India. 4. Acquire the information about the fine arts in India. 5. Know the contribution of scientists of different eras. 							

UNIT - I

Introduction to Culture: Culture, civilization, culture and heritage, general characteristics of culture, importance of culture in human literature, Indian Culture, Ancient India, Medieval India, Modern India

UNIT - II

Indian Languages, Culture and Literature: Indian Languages and Literature-I: the role of Sanskrit, significance of scriptures to current society, Indian philosophies, other Sanskrit literature, literature of south India

Indian Languages and Literature-II: Northern Indian languages & literature

UNIT - III

Religion and Philosophy: Religion and Philosophy in ancient India, Religion and Philosophy in Medieval India, Religious Reform Movements in Modern India (selected movements only)

UNIT – IV

Fine Arts in India (Art, Technology & Engineering): Indian Painting, Indian handicrafts, Music, divisions of Indian classic music, modern Indian music, Dance and Drama, Indian Architecture (ancient, medieval and modern), Science and Technology in India, development of science in ancient, medieval and modern India

UNIT – V

Education System in India: Education in ancient, medieval and modern India, aims of education, subjects, languages, Science and Scientists of Ancient India, Science and Scientists of Medieval India, Scientists of Modern India

Suggested Reading:

1. Kapil Kapoor, "Text and Interpretation: The India Tradition", ISBN: 81246033375, 2005
2. "Science in Samskrit", Samskrita Bharti Publisher, ISBN 13: 978-8187276333, 2007
3. NCERT, "Position paper on Arts, Music, Dance and Theatre", ISBN 81-7450 494-X, 2000

4. S. Narain, "Examinations in ancient India", Arya Book Depot, 1993
5. Satya Prakash, "Founders of Sciences in Ancient India", Vijay Kumar Publisher, 1989
6. M. Hiriyanna, "Essentials of Indian Philosophy", Motilal Banarsidass Publishers, ISBN 13: 978-8120810990, 2014

Course Code	Course Title				Core / Elective		
BS102MT	Mathematics - I (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	3	1	-	-	30	70	4
Course Objectives <ul style="list-style-type: none"> ➤ To introduce the concepts of sequences, series and their properties ➤ To introduce the concepts of functions of several variables and multiple integrals ➤ To study vector differential and integral calculus Course Outcomes The students will able to <ol style="list-style-type: none"> 1. Find the nature of sequences and series 2. Evaluate multiple integrals 3. Apply this knowledge to solve the curriculum problems 							

Unit-I

Sequences and Series: Sequences, Series, General properties of series, Series of positive terms, Comparison tests, tests of Convergence D'Alembert's ratio test, Cauchy's n^{th} root test, Raabe's test, Logarithmic test, Alternating series, Series of positive and negative terms, Absolute convergence and Conditional convergence.

Unit-II:

Calculus of one Variable: Rolle's theorem, Lagrange's, Cauchy's mean value theorems, Taylor's series, Curvature, Radius of curvature, Circle of curvature, Envelope of a family of curves, Evolutes and Involutives.

Unit-III

Multivariable Calculus (Differentiation): Functions of two variables, Limits and continuity, Partial derivatives, Total differential and differentiability, Derivatives of composite and implicit functions (Chain rule), Change of variables, Jacobian, Higher order partial derivatives, Taylor's series of functions of two variables, Maximum and minimum values of functions of two variables, Lagrange's method of undetermined multipliers.

Unit-IV

Multivariable Calculus (Integration): Double integrals, Change of order of integration, Change of Variables from Cartesian to plane polar coordinates, Triple integrals.

Unit-V

Vector Calculus: Scalar and vector fields, Gradient of a scalar field, Directional derivative, Divergence and Curl of a vector field, Line, Surface and Volume integrals, Green's theorem in a plane, Gauss's divergence theorem, Stoke's theorem (without proofs) and their verification.

Suggested Readings:

1. R.K. Jain & S.R.K Iyengar, *Advanced Engineering Mathematics*, Narosa Publications, 2014.
2. Erwin Kreyszig, *Advanced Engineering Mathematics*, John Wiley, 9th Edition, 2012.
3. B.S. Grewal, *Higher Engineering Mathematics*, Khanna Publications, 43rd Edition, 2014.
4. G.B. Thomas, Maurice Weir and Joel Hass, *Thomas' Calculus*, Peterson, 12th Edition, 2010.
5. B.V. Ramana, *Higher Engineering Mathematics*, 23rd reprint, 2015.

Course Code	Course Title				Core / Elective		
BS105CH	Chemistry (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	3	1	-	-	30	70	4
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ Correlate the properties of materials with their internal structure and use the for Engineering applications ➤ Apply the principals of electrochemistry in storage of electrical energy in batteries. ➤ Gains knowledge in causes of corrosion and its prevention. ➤ Attains knowledge about the disadvantages of hard water for domestic and industrial purposes. Also learns the techniques of softening of hard water and treatment of water for drinking purpose. ➤ Exposed to qualitative and quantitative parameters of chemical fuels. ➤ Aware eco-friendly materials and processes. <p>Course Outcomes</p> <p>On successful completion of this course, students will be able to:</p> <ol style="list-style-type: none"> 1. Apply concept of electrode potential in identifying feasibility of electrochemical reaction; illustrate electro analytical techniques and working of batteries. 2. Identify the mechanism of corrosion of materials on basis of electrochemical approach and devise corrosion control methods. 3. Estimate the physical & chemical parameters of quality of water and explain the process of water treatment. 4. Explain the influence of chemical structure on properties of materials and their choice in engineering applications. 5. Classify chemical fuels and grade them through qualitative analysis. 6. Relate the concept of green chemistry to modify engineering processes and materials. 							

UNIT-I

Electrochemistry and Battery Chemistry: Electrochemistry: Electrochemical cells, Electrolytic and Galvanic cells-notation, cell reaction and cell potentials. Types of electrodes, Calomel Quinhydrone and Glass electrodes. Determination of pH of a solution by using Quinhydrone electrode. Thermodynamics of emf of cells, Nernst equation and its derivation. Applications of Nernst equation to electrode potential and emf of cells. Numerical problems.

Batteries: Primary batteries: Zn - Carbon battery. **Secondary batteries:** Pb-Acid battery and Li-Ion battery, Applications. **Flow batteries (Fuel cells):** Methanol-Oxygen fuel cells, Construction, Applications.

UNIT-II

Water Chemistry and Corrosion: Water Chemistry: Hardness of Water-Types and units of hardness, estimation of temporary and permanent hardness of water by EDTA method. Alkalinity of water and its determination. Water softening by Ion exchange and Reverse Osmosis methods. Numerical problems. Specifications of potable water. Sterilization by Chlorination. Break Point Chlorination.

Corrosion: Causes and its effects. Types of Corrosion-Dry or Chemical corrosion and Wet or Electrochemical corrosion and their mechanism. Electrochemical corrosion –Waterline and Pitting Corrosion. Factors influencing rate of corrosion.

Corrosion control methods: Cathodic protection methods - Sacrificial anodic and impressed current methods. Surface coating methods: Hot Dipping-Galvanizing.

UNIT-III

Engineering Materials: Polymers: Basics of terms polymers: Monomer and its functionality, Polymers and degree of polymerization. Classification of polymers - Thermoplastics & Thermosetting resins.

Types of Polymerization (i) Addition (ii) Condensation (iii) Co-Polymerization. Mechanism of free radical polymerization

Preparation, Properties & Uses of the following polymers: Plastics - PVC and Bakelite, Fibres - Nylon 6:6, and Kevlar, Elastomers - Buna-S, Butyl and Silicone Rubbers.

Conducting polymers : Introduction, Classification and Mechanism of conduction in Poly-acetylene, Applications of conducting polymers.

Biodegradable polymers: Introduction preparation, properties and applications of polylactic acid

UNIT-IV

Chemical Fuels: Classification of fuels: Introduction, definition and classification of chemical fuels- Primary and secondary fuels. Solid, liquid and gaseous fuels. Requirements of a good fuel. Calorific Value – HCV and LCV. Theoretical calculations of calorific value by Dulong's formula – Numerical problems.

Solid Fuels: Coal and its Ranking. Analysis of coal - Proximate and Ultimate analysis.

Liquid Fuels: Fractionation of Petroleum. Composition and uses of Gasoline, Diesel and Kerosene. Cracking & its Significance- Catalytic cracking by moving bed method, Knocking. Fuel rating – Octane and Cetane numbers.

Gaseous Fuels: LPG, CNG -Composition and Uses.

Combustion: Ignition temperature of a fuel, calculation of air quantities by weight and volume required for combustion of a fuel- Numerical problems.

UNIT-V

Green Chemistry and Composites: Green Chemistry: Concept, Principles of green chemistry – Atom Economy, Catalysis. and examples of clean technology.

Biodiesel: Sources, Concept of Trans esterification and carbon neutrality. Properties and significance

Composites: Introduction to composites, composition and characteristic properties of composites. Classification of composites based on matrix, reinforcement and ply. Applications of composites.

Suggested Readings:

1. Principles of Physical Chemistry by Puri, Sharma and Pathania S.N. Chand & Co. New Delhi (Latest edition).
2. Engineering Chemistry by P C Jain and M Jain Dhanpat Rai & Sons (15th Edn), New Delhi.
3. Chemistry in Engineering and Technology by J C Kuriacose and J Rajaram, TMH, New Delhi.
4. Engineering Chemistry by O G Palanna, TMH, and New Delhi.
5. Engineering Chemistry by S S Dara, S Chand & Sons, New Delhi.
6. Engineering Chemistry by Sashi Chawla. Dhanpat Rai & Sons, New Delhi.
7. Engineering Chemistry by Shikha Agrawal, Cambridge, New Delhi.
8. Engineering Chemistry by Prasanta Rath, Cengage Learning India Pvt. Ltd.

Course Code	Course Title				Core / Elective		
ES107CS	Programming for Problem Solving (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	3	-	-	-	30	70	3
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ To introduce the basic concepts of Computing environment, number systems and flowcharts ➤ To familiarize the basic constructs of C language – data types, operators and expressions ➤ To understand modular and structured programming constructs in C ➤ To learn the usage of structured data types and memory management using pointers ➤ To learn the concepts of data handling using pointers <p>Course Outcomes The students will able to</p> <ol style="list-style-type: none"> 1. Formulate simple algorithms for arithmetic and logical problems. 2. Translate the algorithms to programs (in c language). 3. Test and execute the programs and correct syntax and logical errors. 4. Implement conditional branching, iteration and recursion. 5. Decompose a problem into functions and synthesize a complete program using divide and conquer approach. 6. Use arrays, pointers and structures to formulate algorithms and programs. 7. Apply programming to solve matrix addition and multiplication problems and searching and sorting problems. 8. Apply programming to solve simple numerical method problems, namely rot finding of function, differentiation of function and simple integration. 							

Unit - I

Introduction to Programming: Introduction to components of a computer system (disks, memory, processor, where a program is stored and executed, operating system, compilers etc.).

Idea of Algorithm: steps to solve logical and numerical problems.

Representation of Algorithm: Flowchart / Pseudocode with examples. From algorithms to programs; source code, variables (with data types) variables and memory locations, Syntax and Logical Errors in compilation, object and executable code.

Unit - II

Control Structures: Arithmetic expressions and precedence, Conditional Branching and Loops, Writing and evaluation of conditionals and consequent branching.

Arrays: Arrays (1-D, 2-D), Character arrays and Strings

Unit - III

Basic Algorithms: Searching, Basic Sorting Algorithms (Bubble and Selection), Finding roots of Equations. **Functions:** Functions (including using built in libraries), Parameter passing in functions, call by value. **Passing arrays to functions:** idea of call by reference

Unit - IV

Recursion: Recursion, as a different way of solving problems. Example programs, such as Finding Factorial, Fibonacci series. **Structure:** Structures, Defining structures and Array of Structures

Unit - V

Pointers - Idea of pointers, Defining pointers, Use of Pointers in self-referential structures, notion of linked list (no implementation), **Introduction to File Handling.**

Suggested Readings:

1. Byron Gottfried, Schism's Outline of Programming with C, McGraw-Hill
2. A.K. Sharma, Computer Fundamentals and Programming in C, Universities Press, 2nd Edition, 2018.
3. E. Balaguruswamy, Programming in ANSI C, Tata McGraw-Hill
4. Brian W. Kernighan and Dennis M. Ritchie, the C Programming Language, Prentice Hall of India.

Course Code	Course Title					Core / Elective	
BS 153 CH	Chemistry Lab (Common to All Branches)					Core	
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	-	-	-	3	25	50	1.5
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ Conduct experiments, take measurements and analyse the data through hands-on experience in order to demonstrate understanding of the theoretical concepts of quantitative Analysis while working in small group. ➤ Interpret the electro analytical principles with experimental results graphically ➤ Demonstrate writing skills through clear laboratory reports <p>Course Outcomes</p> <p>On successful completion of this course, students will be able to:</p> <ol style="list-style-type: none"> 1. Apply the principles of Colourimetry and Electrochemistry in quantitative estimations. 2. Estimate the rate constants of reactions from concentration of reactants/ products as a function of time. 3. Synthesize small drug molecules. 							

List of Experiments:

1. Introduction to Chemical Analysis.
2. Techniques of Weighing.
- Volumetric Analysis:**
 3. Preparation of Standard Mohr's salt solution, Standardization of KMnO_4 and estimation ferrous ion.
 4. Estimation Iron(II) by Dichromatometry
- Water Analysis:**
 5. Preparation of Standard Magnesium sulphate solution, standardization of EDTA and Estimation of Total Hardness.
 6. Preparation of Standard Sodium Carbonate Solution, Standardization of HCl and Estimation of Carbonate and Bicarbonate Alkalinity.
- Conductometry:**
 7. Estimation of HCl
 8. Estimation of CH_3COOH and mixture of acids
- Potentiometry**
 9. Estimation of HCl
 10. Estimation of Iron
- pH Metry:**
 11. Estimation of HCl
- Colorimetry:**
 12. Estimation of HCl
 13. Verification of Beer-Lambert's law and estimation of Manganese.
- Chemical Kinetics:**
 14. Determination of rate constant of acid catalysed hydrolysis of methyl acetate.
- 16. Drug Synthesis**
 15. Preparation of Aspirin

Note: Minimum ten experiments should be conducted in the semester

Suggested Readings:

1. Senior Practical Physical Chemistry, B.D. Khosla, A. Gulati and V. Garg (R. Chand & Co., Delhi)
2. An Introduction to Practical Chemistry, K. K. Sharma and D.S. Sharma (Vikas publishing, N. Delhi)

Course Code	Course Title				Core / Elective		
ES 155 CS	Programming for Problem Solving Lab (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	-	-	-	4	25	50	2
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ Understand the fundamentals of programming in C Language. ➤ Write, compile and debug programs in C. ➤ Formulate solution to problems and implement in C. ➤ Effectively choose programming components to solve computing problems <p>Course Outcomes <i>The students will able to</i></p> <ol style="list-style-type: none"> 1. Choose appropriate data type for implementing programs in C language. 2. Design and implement modular programs involving input output operations, decision making and looping constructs. 3. Implement search and sort operations on arrays. 4. Apply the concept of pointers for implementing programs on dynamic memory management and string handling. 5. Design and implement programs to store data in structures and files. 							

Programming Exercise:

1. Finding maximum and minimum of given set of numbers, finding roots of quadratic equation.
2. Sin x and Cos x values using series expansion.
3. Conversion of binary to decimal, octal, hexadecimal and vice versa.
4. Generating Pascal triangle, pyramid of numbers.
5. Recursion: factorial, Fibonacci, GCD.
6. Matrix addition and multiplication using arrays, linear search and binary search using recursive and non-recursive procedures.
7. Bubble sort and selection sort.
8. Programs on pointers: pointer to arrays, pointer to functions.
9. Functions for string manipulations.
10. Programs on structures and unions.
11. Finding the number of characters, words and lines of given text file.
12. File handling programs

Suggested Readings:

1. Byron Gottfried, Schaum's Outline of Programming with C, McGraw-Hill
2. A.K. Sharma, Computer Fundamentals and Programming in C, Universities Press, 2018.
3. E. Balaguruswamy, Programming in ANSI C, Tata McGraw-Hill
4. Brian W. Kernighan and Dennis M. Ritchie, the C Programming Language, Prentice Hall of India.

Course Code	Course Title				Core / Elective		
ES 157 ME	Workshop/ Manufacturing Process (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	1	-	-	4	50	50	3
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ Identify and use marking out tools, hand tools, measuring equipment and to work to prescribed tolerances. ➤ To provide hands on experience about use of different engineering materials, tools, equipments and processes those are common in the engineering field. ➤ To gain a good basic working knowledge required for the production of various engineering products. ➤ To Study different hand operated power tools, uses and their demonstration. ➤ Adopt safety practices while working with various tools <p>Course Outcomes The students will able to</p> <ol style="list-style-type: none"> 1. Demonstrate an understanding of and comply with workshop safety regulations. 2. Identify and apply suitable tools for different trades of Engineering processes including drilling, material removing, measuring, chiselling. 3. Study and practice on machine tools and their operations 4. Undertake jobs connected with Engineering Workshop trades including fitting, carpentry, sheet metal, house wiring, welding, smithy and foundry. 5. Apply basic electrical engineering knowledge for house wiring practice 							

A. TRADE FOR EXERCISES:

1. Carpentry
2. Fitting
3. House wiring
4. Sheet metal working
5. Smithy
6. Welding
7. Plumbing

B. TRADES FOR DEMONSTRATION AND EXPOSURE:

1. Machining (Lathe & Drilling)
2. Injection moulding
3. Mould making and casting
4. Basic Electronics lab instruments

C. PRESENTATIONS AND VIDEO LECTURES

1. Manufacturing Methods
2. Rapid Prototyping
3. Glass Cutting
4. 3D printing
5. CNC LATHE

- D. IT WORKSHOP:** Computer hardware, identification of parts, Disassembly, Assembly of computer to working condition, operating system installation.

Suggested Reading:

1. Venugopal, K, "Workshop manual", Anuradha Publications, Kumbakonam, TN, 2012
2. K.C. John, "Mechanical Workshop" 2nd Edn., PHI, 2010.
3. Hajra Choudary, "Elements of Workshop Technology" Vol. 1, Asian Publishers, Edn., 1993.
4. G.S. Sawhney, "Mechanical Experiments and Workshop Practice", I.K. International Publishing House, New Delhi, 2009.

Note: At least two exercises from each trade.

SCHEME OF INSTRUCTION & EXAMINATION**B.E. (All Branches) II - Semester****(Group B – CSE, CE, EEE, EIE)**

S. No.	Course Code	Course Title	Scheme of Instructions				Scheme of Examination			Credits
			L	T	P/D	Contact Hours/Week	CIE	SEE	Duration in Hours	
Theory Course										
1	MC111PO	Indian Constitution	2	-	-	2	30	70	3	-
2	HS101EG	English	2	-	-	2	30	70	3	2
3	BS103MT	Mathematics-II	3	1	-	4	30	70	3	4
4	BS104PH	Physics	3	1	-	4	30	70	3	4
5	ES106EE	Basic Electrical Engineering	3	1	-	4	30	70	3	4
Practical/ Laboratory Course										
6	HS151EG	English Lab	-	-	2	2	25	50	3	1
7	BS152PH	Physics Lab	-	-	3	3	25	50	3	1.5
8	ES154EE	Basic Electrical Engineering Lab	-	-	2	2	25	50	3	1
9	ES156CE	Engineering Graphics & Design	1	-	4	5	50	50	3	3
		Total	14	03	11	28	275	550		20.5

HS: Humanities and Social Sciences **BS:** Basic Science **ES:** Engineering Science
MC: Mandatory Course

L: Lectures **T:** Tutorials **P:** Practical **D:** Drawing
CIE: Continuous Internal Evaluation **SEE:** Semester End Examination (Univ. Exam)

PO: Political Science **EG:** English **MT:** Mathematics **PH:** Physics
EE: Electrical Engineering **CE:** Civil Engineering

Note:

- Each contact hour is a Clock Hour.
- The students have to undergo a Summer Internship of Rural Agriculture Work Experience (**RAWE**) of one week duration after II-Semester and credits will be awarded in VII semester after evaluation.
- Rural Agriculture Work Experience helps the students primarily to understand the rural situations, status of Agricultural Technologies adopted by farmers and village development plans and to develop skills & attitude of working with farm families for overall development in rural area.
- The main objectives of RAWE component are:
 - To make the students familiar with socio-economic conditions of the farmers.
 - To develop communication skills in students using extension teaching methods in transfer of Technology.

Course Code	Course Title					Core/Elective	
MC111PO	Indian Constitution (Common to All Branches)					Mandatory Course	
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	2	-	-	-	30	70	-
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ To create awareness among students about the Indian Constitution. ➤ To acquaint the working conditions of union, state, local levels, their powers and functions. ➤ To create consciousness in the students on democratic values and principles articulated in the constitution. ➤ To expose the students on the relations between federal and provincial units. ➤ To divulge the students about the statutory institutions. <p>Course Outcomes</p> <p>After completing this course, the student will</p> <ol style="list-style-type: none"> 1. Know the background of the present constitution of India. 2. Understand the working of the union, state and local levels. 3. Gain consciousness on the fundamental rights and duties. 4. Be able to understand the functioning and distribution of financial resources between the centre and states. 5. Be exposed to the reality of hierarchical Indian social structure and the ways the grievances of the deprived sections can be addressed to raise human dignity in a democratic way. 							

UNIT-I

Evolution of the Indian Constitution: 1909 Act, 1919 Act and 1935 Act. Constituent Assembly: Composition and Functions; Fundamental features of the Indian Constitution.

UNIT-II

Union Government: Executive-President, Prime Minister, Council of Minister

State Government: Executive: Governor, Chief Minister, Council of Minister

Local Government: Panchayat Raj Institutions, Urban Government

UNIT-III

Rights and Duties: Fundamental Rights, Directive principles, Fundamental Duties

UNIT-IV

Relation between Federal and Provincial units: Union-State relations, Administrative, legislative and Financial, Inter State council, NITI Ayog, Finance Commission of India

UNIT-V

Statutory Institutions: Elections-Election Commission of India, National Human Rights Commission, National Commission for Women

Suggested Readings:

1. D.D. Basu, Introduction to the constitution of India, Lexis Nexis, New Delhi
2. Subhash Kashyap, Our Parliament, National Book Trust, New Delhi
3. Peu Ghosh, Indian Government & Politics, Prentice Hall of India, New Delhi
4. B.Z. Fadia & Kuldeep Fadia, Indian Government & Politics, Lexis Nexis, New Delhi

Course Code	Course Title				Core / Elective		
HS 101 EG	English (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	2	-	-	-	30	70	2
<p>Course Objectives: To enhance the English language abilities of Engineering students especially in reading and writing, by</p> <ul style="list-style-type: none"> ➤ Using authentic material for language learning ➤ Exposing them to a variety of content-rich texts ➤ Strengthening their grammar and vocabulary ➤ Improving their reading and comprehension skills ➤ Honing their writing skills ➤ Encouraging them to think creatively and critically <p>Course Outcomes: On successful completion of the course, the student will be able to</p> <ol style="list-style-type: none"> 1. Read, understand, and interpret a variety of written texts 2. Use appropriate vocabulary and correct grammar 3. Undertake guided and extended writing with confidence. 							

Unit – I**Reading:** RK Narayan, “A Horse and Two Goats”**Vocabulary:** Word formation—Prefixes, Suffixes, Root Words**Grammar:** Articles, Prepositions, Determiners**Unit – II****Reading:** Rudyard Kipling, “If”**Vocabulary:** Word formation—Compounding and Blending, Contractions**Grammar:** Transitions, Connectives**Writing:** Paragraph Writing**Unit – III****Reading:** Martin Luther King Jr., “I Have a dream”**Vocabulary:** Synonyms, Antonyms, One Word Substitutes**Grammar:** Voice**Writing:** Letter Writing**Unit – IV****Reading:** Robert Frost, “Road Not Taken”**Vocabulary:** Homophones, Homonyms, Homographs**Grammar:** Narration (Direct-Indirect Speech)**Writing:** Report Writing**Unit – V****Reading:** George Orwell, “The Sporting Spirit” (Excerpt)**Vocabulary:** Inclusive Language, Euphemisms**Grammar:** Tense**Writing:** SOP**Suggested Readings:**

1. Board of Editors. Language and Life: A Skills Approach. Orient Black Swan, 2018.
2. Sudharshana, NP and C Savitha. English for Engineers. Cambridge University Press, 2018.
3. Kumar, Sanjay and Pushp Lata. English Language and Communication Skills for Engineers, Oxford University Press, 2018.

Course Code	Course Title				Core / Elective		
BS 103 MT	Mathematics – II (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	3	1	-	-	30	70	4
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ To study matrix algebra and its use in solving system of linear equations and in solving eigen value problems ➤ To provide an overview of ordinary differential equations ➤ To study special functions like Legendre and Beta Gamma functions ➤ To learn Laplace Transforms and its properties <p>Course Outcomes The students will able to</p> <ol style="list-style-type: none"> 1. Solve system of linear equations and eigen value problems 2. Solve certain first order and higher order differential equations 3. Solve basic problems of Beta Gamma and Legendre's Function. 4. Apply Laplace Transforms; solve ordinary Differential Equations by using it. 							

Unit-I

Matrices: Rank of a matrix, Echelon form, System of linear equations, Linearly dependence and independence of vectors, Linear transformation, Orthogonal transformation, Eigen values, Eigenvectors, Properties of eigen values, Cayley - Hamilton theorem, Quadratic forms, Reduction of quadratic form to canonical form by orthogonal transformation, Nature of quadratic forms.

Unit-II

Differential Equations of First Order: Exact differential equations, Integrating factors, Linear differential equations, Bernoulli's, Riccati's and Clairaut's differential equations, Orthogonal trajectories of a given family of curves.

Unit-III

Differential Equations of Higher Orders: Solutions of second and higher order linear homogeneous equations with constants coefficients, Method of reduction of order for the linear homogeneous second order differential equations with variable coefficients, Solutions of non-homogeneous linear differential equations, Method of variation of parameters, solution of Euler-Cauchy equation.

Unit-IV

Special Function: Gamma Functions, Beta Functions, Relation Between Beta and Gamma Function, Error Functions. Power Series Method, Legendre's Differential Equations and Legendre's Polynomial $P_n(x)$, Rodrigue's Formula (without proof).

Unit-V

Laplace Transforms: Laplace Transforms, Inverse Laplace Transforms, Properties of Laplace Transforms and inverse Laplace Transforms, Convolution Theorem (without proof). Solution of ordinary Differential Equations using Laplace Transforms.

Suggested Readings:

1. R.K. Jain & S.R.K. Iyengar, *Advanced Engineering Mathematics*, Narosa Publications, 4th Edition, 2014.
2. Erwin Kreyszig, *Advanced Engineering Mathematics*, John Wiley, 9th Edition, 2012.
3. Dr.B.S. Grewal, *Higher Engineering Mathematics*, Khanna Publications, 43rd Edition, 2014.

4. B.V. Ramana, *Higher Engineering Mathematics*, 23rd reprint, 2015.
5. N. Bali, M. Goyal, A text book of Engineering *Mathematics*, Laxmi publications, 2010
6. H.K. Dass, Er. Rajnish Varma, *Higher Engineering Mathematics*, Schand Technical Third Edition.

Course Code	Course Title				Core / Elective		
BS 104 PH	Physics (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	3	1	-	-	30	70	4
Course Objectives <ul style="list-style-type: none"> ➤ Aware of limits of classical free electron free theory and to apply band theory of solids ➤ Acquire knowledge on various properties of semiconductors. ➤ Grasp the intricacies in semiconductor-optical interaction Course Outcomes <ol style="list-style-type: none"> 1. Distinguish materials based on band theory of solids 2. Classify semiconductors on the basis doping and to estimate conductivity and learn transport phenomenon in semiconductors 3. Appreciate use of optical absorption by semiconductors. 							

Unit – I

Crystallography: Introduction, Types of crystal systems, Bravais lattices, Lattice planes and Miller Indices (Cubic system), Inter planar spacing (Cubic system), Bragg's law, Powder diffraction method.

Crystal Defects: Classification of point defects, Concentration of Schottky defects in metals and ionic crystals, Concentration of Frankel defects, Line defects, Screw and Edge dislocations, Burger's vector

Unit – II

Band Theory of Solids & Semiconductors: Classical free electron theory (qualitative), Kronig Penney model (qualitative treatment), Energy band formation in solids, Intrinsic and Extrinsic semiconductors, Concept of a hole, Carrier concentration and conductivity in intrinsic semiconductors, Formation of P-N junction diode and its I – V characteristics, Thermistor and its characteristics, Hall effect and its applications.

Dielectric Materials: Dielectrics, Types of polarizations, Electronic, Ionic, Orientational and Space charge polarizations, Expression for Electronic polarizability, Frequency and temperature dependence of dielectric polarizations, Determination of dielectric constant by capacitance Bridge method, Ferroelectricity, Barium titanate, Applications of Ferroelectrics.

Unit – III

Wave Mechanics: Matter waves –de-Broglie wavelength, properties of wave function, Physical significance, Schrodinger time dependent and time in-dependent wave equation. Particle in a 1-D box.

Electromagnetic theory: Basic laws of electricity and magnetism, Maxwell's equations in integral and differential forms, Conduction and displacement current, Relation between D, E and P –

Electromagnetic waves: Equation of plane wave in free space, Poynting theorem.

Unit – IV

Magnetic Materials: Classification of magnetic materials: dia, para, ferro, antiferro and ferrimagnetic materials, Weiss molecular field theory of ferromagnetism, Magnetic domains, Hysteresis curve, soft and hard magnetic materials, Ferrites: Applications of ferrites.

Superconductivity: Introduction, General properties of super conductors, Meissner effect, Type I and Type II superconductors, BCS theory (qualitative), Introduction to High T_c superconductors, Applications of superconductors.

Unit – V

Lasers: Characteristics of Lasers, spontaneous and stimulated emission of radiation, Einstein's Coefficients, population inversion, Ruby Laser, Helium Neon Laser, Semi-Conductor Laser and applications of lasers.

Fiber Optics: Introduction, Propagation of light through an optical fiber, Acceptance angle, Numerical aperture (NA), Types of Optical fibers and Refractive index profiles, Fiber drawing process (double Crucible Method), Losses in optical fibers, applications of optical fibers.

Suggested Reading:

1. B.K. Pandey and S. Chaturvedi Engineering Physics Cengage Learning 2012
2. A.K. Bhandhopadhyaya, Nano Materials, New Age International, 1st Edition, 2007
3. M.S. Avadhanulu and P.G. Kshirusagar, Engg. Physics, S. Chand & Co. 1st Edition, 1992.
4. C.M. Srivastava and C. Srinivasan – Science of Engg Materials, New Age International.
5. R.K Gaur and S.L Gupta- Engineering Physics, Dhanpathrai Publications, New edition.
6. Sanjay D Jain & Girish G Sahasrabudhe -Engineering Physics, University Press

Course Code	Course Title				Core / Elective		
ES 106 EE	Basic Electrical Engineering (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	3	1	-	-	30	70	4
Course Objectives <ul style="list-style-type: none"> ➤ To provide an understanding of basics in Electrical circuits. ➤ To explain the working principles of Electrical Machines and single phase transformers. Course Outcomes <ol style="list-style-type: none"> 1. To analyse Electrical circuits to compute and measure the parameters of Electrical Energy. 2. To comprehend the working principles of Electrical DC Machines. 3. To Identify and test various Electrical switchgear, single phase transformers and assess the ratings needed in given application. 4. To comprehend the working principles of electrical AC machines. 							

Unit-I

DC Circuits: Electrical circuit elements (R, L and C), voltage and current sources, Kirchoff current and voltage laws, analysis of simple circuits with dc excitation. Superposition, Thevenin and Norton Theorems.

Unit-II

AC Circuits: Representation of sinusoidal waveforms, peak and rms values, phasor representation, real power, reactive power, apparent power, power factor. Analysis of single-phase ac circuits consisting of R, L, C, and RL, RC, RLC combinations (series only). Three phase balanced circuits, voltage and current relations in star and delta connections.

Unit-III

Transformers and 3-ph Induction Motors: Transformers: Electromagnetic induction, Faradays laws, statically induced emf, Lenz law, BH characteristics, ideal and practical transformer, losses and efficiency, Auto-transformer and three-phase transformer connections.

Three Phase Induction motor: Generation of rotating magnetic fields, Construction and working of a three-phase induction motor, squirrel cage IM, slip-ring IM, Applications.

Unit-IV

Single-phase induction motor & DC Machines: Single-phase induction motor: Construction and principle of operation, Capacitor start & capacitor run motor, applications

DC Generators: Dynamically induced emf, Flemming's Right hand and Left hand rules, Construction and principle of operation of DC generator, EMF equation, Types of DC Generators, OCC characteristics, applications

DC Motors: principle of operation of DC Motor, Types of DC motors, applications.

Unit-V

Electrical Installations: Components of LT Switchgear: Switch Fuse Unit (SFU), MCB, ELCB, MCCB, Types of Wires and Cables, Earthing. Types of Batteries, Important Characteristics for Batteries. Elementary calculations for energy consumption, power factor improvement and battery backup.

Suggested Reading:

1. N.K. De, "Basic Electrical Engineering", Universities Press, 2015.
2. J.B. Gupta, "Fundamentals of Electrical Engineering and Electronics" S.K. Kataria & Sons Publications, 2002.

3. J.B. Gupta, "Utilization of Electric Power and Electric Traction" S.K. Kataria & Sons Publications, 2010
4. Abhijit Chakrabarti, Sudipta Nath, Chandan Kumar Chanda, "Basic Electrical Engineering" Tata McGraw Hill, Publications, 2009
5. Hughes, "Electrical Technology", VII Edition, International Student -on, Addison Welsey Longman Inc., 1995.

Course Code	Course Title				Core / Elective		
HS 151 EG	English Lab (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	-	-	-	2	25	50	1
<p>Course Objectives To enhance the listening and speaking skills of students by</p> <ul style="list-style-type: none"> ➤ Giving them sufficient practice in listening with comprehension ➤ Providing them ample opportunities to improve their public speaking skills ➤ Training them in the use of correct pronunciation, stress, and intonation ➤ Sensitizing them to the use of verbal and non-verbal communication appropriate to the context ➤ Encouraging them to learn the art of conversation to suit formal and informal situations ➤ Preparing them to make formal presentations and face interviews <p>Course Outcomes On successful completion of the course, students will be able to</p> <ol style="list-style-type: none"> 1. Listen, understand, and interpret formal and informal spoken language 2. Speak English with acceptable pronunciation, stress, and intonation 3. Present themselves with confidence in formal situations 4. Participate in individual and group activities with relative ease 							

List of Experiments:

1. Listening for Comprehension
2. Pronunciation, Intonation, Stress, and Rhythm
3. Conversation Skills
4. Introducing Oneself and Others
5. Asking for and Giving Information
6. Making Requests and Responding to them Appropriately
7. Giving Instructions and Responding to them Appropriately
8. Making Formal Announcements and Emceeing
9. Group Discussions
10. JAM
11. Role Play
12. Debate
13. Public Speaking Skills and Body Language
14. Interviews
15. Formal Presentations

Suggested Readings:

1. Board of Editors. Language and Life: A Skills Approach. Orient Black Swan, 2018.
2. Balasubramanian, T. A Textbook of English Phonetics for Indian Students. Macmillan, 1981.
3. CIEFL. Exercises in Spoken English. Parts. I-III. Oxford University Press.
4. Pillai, Radhakrishna G. Spoken English For You - Level II. 8th Edition. Emerald Publishers, 2014.
5. Sethi, J and PV Dhamija. A Course in Phonetics and Spoken English. 2nd Edition, Prentice Hall India Learning Private Limited, 1999.

Course Code	Course Title					Core / Elective	
BS 152 PH	Physics Lab (Common to All Branches)					Core	
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	-	-	-	3	25	50	1.5
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ Make precise measurements using basic physical principles and acquire skills to handle the instruments ➤ Relates the theoretical Knowledge to the behavior of Practical Physical world. ➤ Analyse errors in the experimental data. ➤ Plot graphs between various physical parameters. <p>Course Outcomes</p> <ol style="list-style-type: none"> 1. Conduct experiments, take measurements independently. 2. Write appropriate laboratory reports. 3. Compute and compare the experimental results and draw relevant conclusions. 4. Use the graphical representation of data and estimate results from graphs 							

List of Experiments:

1. To determine the Dielectric constant and Phase transition temperature of Lead Zirconium Titanate (PZT).
2. To draw the I - V Characteristics of P-N Junction diode and to evaluate the resistance.
3. To find the values of Electrical conductivity and energy gap of Ge crystal.
4. Determination of rigidity of modulus of Torsion pendulum.
5. Determination of carrier concentration, Mobility and Hall Coefficient of Ge crystal using Hall Effect Experiment.
6. To determine the constants of A, B and α using Thermistor characteristics.
7. To draw the curve between the magnetizing field and the intensity of magnetization of the specimen (soft iron rod) and to find out
 - i) Coercivity ii) Retentivity and iii) Hysteresis loss.
8. To draw the I - V Characteristics of a solar cell and to calculate the
 - i) Fill factor Efficiency and ii) Series resistance.
9. To Determine the Numerical aperture (NA) of Optical fiber.
10. To determine the wave length of the given Laser source.

Note: Minimum eight experiments should be conducted in the semester

Suggested Reading:

1. N.K. De, "Basic Electrical Engineering", Universities Press, 2015.
2. J.B. Gupta, "Fundamentals of Electrical Engineering and Electronics" S.K. Kataria & Sons Publications, 2002.
3. J.B. Gupta, "Utilization of Electric Power and Electric Traction" S.K. Kataria & Sons Publications, 2010

Course Code	Course Title					Core / Elective	
ES 154 EE	Basic Electrical Engineering Lab (Common to All Branches)					Core	
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	-	-	-	2	25	50	1
<p>Course Objectives</p> <ul style="list-style-type: none"> ➤ To impart the practical knowledge on testing of DC and AC Machines and the usage of common electrical measuring instruments <p>Course Outcomes</p> <ol style="list-style-type: none"> 1. Get an exposure to common electrical components and their ratings. 2. Analyse the performance of DC and AC Machines. 3. Comprehend the usage of common electrical measuring instruments. 4. Test the basic characteristics of transformers and electrical machines. 							

Suggested List of Laboratory Experiments/Demonstrations:

- Dem1. Basic safety precautions. Introduction and use of measuring instruments – voltmeter, ammeter, multi-meter, oscilloscope. Real-life resistors, capacitors and inductors.
- Exp 1. Verification of KVL and KCL, superposition theorem (with DC excitation)
- Exp 2 Verification of Thevenins and Nortons theorems (with DC excitation)
- Exp 3. Sinusoidal steady state response of R-L, and R-C circuits – impedance calculation and verification. Observation of phase differences between current and voltage. Power factor calculation
- Exp 4. Transformers: Observation of the no-load current waveform on an oscilloscope (nonsinusoidal wave-shape due to B-H curve nonlinearity should be shown along with a discussion about harmonics).
- Exp 5. Loading of a transformer: measurement of primary and secondary voltages and currents, and power.
- Exp 6. Three-phase transformers: Star and Delta connections. Voltage and Current relationships (line-line voltage, phase-to-neutral voltage, line and phase currents).
- Exp 7. Measurement of phase voltage/current, line voltage/current and power in a balanced three-phase circuit connected in star and delta
- Dem2. Demonstration of cut-out sections of machines: dc machine (commutator-brush arrangement), induction machine (squirrel cage rotor), synchronous machine (field winding - slip ring arrangement) and single-phase induction machine.
- Exp 8. OCC characteristics of DC Generator
- Exp 9. Synchronous speed of two and four-pole, three-phase induction motors. Direction reversal by change of phase-sequence of connections.
- Exp 10. Power factor improvement of Induction Motor using static capacitors
- Exp 11. Load Test of DC Motor

Note - 1:

- (i) List of Experiments and Demonstrations suggested above are already available in the Laboratory of the electrical department. No need to purchase any extra equipment except Demonstration2 equipments
- (ii) Procurement of Demonstration 2 equipments can be done during the course work of that semester. It can be included in the laboratory.

Note - 2:

- (i) Experiments 9, 10 and Demonstration 3 can be incorporated in the Lab syllabus if the topics concerned to the above experiments are considered in new BEE syllabus.

Suggested Reading:

1. J.B. Gupta, "Fundamentals of Electrical Engineering and Electronics" S.K. Kataria & Sons Publications, 2002.
2. J.B. Gupta, "Utilization of Electric Power and Electric Traction" S.K. Kataria & Sons Publications, 2010
3. Abhijit Chakrabarti, Sudipta Nath, Chandan Kumar Chanda, "Basic Electrical Engineering" Tata McGraw Hill, Publications, 2009
4. Hughes, "Electrical Technology", VII Edition, International Student -on, Addison Welsey Longman Inc., 1995.

Course Code	Course Title				Core / Elective		
ES 156 CE	Engineering Graphics & Design (Common to All Branches)				Core		
Prerequisite	Contact Hours per Week				CIE	SEE	Credits
	L	T	D	P			
-	1	-	4	-	50	50	3
Course Objectives <ul style="list-style-type: none"> ➤ To prepare you to design a system, component, or process to meet desired needs within realistic constraints such as economic, environmental, social, political, ethical, health and safety, manufacturability, and sustainability ➤ To prepare you to communicate effectively ➤ To prepare you to use the techniques, skills, and modern engineering tools necessary for engineering practice. Course Outcomes The students will able to <ol style="list-style-type: none"> 1. Introduction to engineering design and its place in society 2. Exposure to the visual aspects of engineering design 3. Exposure to engineering graphics standards 4. Exposure to solid modelling 5. Exposure to computer-aided geometric design 6. Exposure to creating working drawings 7. Exposure to engineering communication 							

Sheet No	Description of the Topic	Contact Hours	
		Lecture	Drawing
1	Principles of Engineering Graphics and their significance, usage of drawing instruments.	1	
2	Conic Sections – I Construction of ellipse, parabola and hyperbola given focus and eccentricity.	1	2
3	Conic Sections – II Construction of ellipse (given major and minor axis), parabola (given base and height), rectangular hyperbola.		2
4	Cycloids (cycloid & epicycloid)	1	2
5	Involutes (involute of triangle, square & circle)		2
6	Scales (plain & diagonal scales)	1	2 + 2
7	Introduction to AutoCAD Basic commands and simple drawings.		2 + 2
8	Orthographic Projection Projections of points situated in different quadrants.	1	2
9	Projections of straight lines – I Line parallel to both the reference planes, line perpendicular or inclined to one reference plane.	1	2
10	Projections of straight lines – II Line inclined to both the reference planes.	1	2
11	Projections of planes – I Perpendicular planes	1	2

12	Projections of planes – II Oblique planes		2
13	Projections of solids – I Polyhedra and solids of revolution, Projections of solids in simple position.	1	2
14	Projection of solids – II Projections of solids when the axes inclined to one or both the reference planes.	1	2 + 2
15	Section of solids – I When the sectional plane is parallel or perpendicular to one reference plane.	1	2
16	Section of solids – II When the sectional plane is inclined to one reference plane.		2
17	Development of surfaces – I Prisms and Cylinders	1	2
18	Development of surfaces – II Pyramids and Cones		2
19	Intersection of surfaces – I Intersection of cylinder and cylinder	1	2
20	Intersection of surfaces – II Intersection of cylinder and cone		2
21	Isometric projection – I planes and simple solids	1	2
22	Isometric projection – II combination of two or three solids		2
23	Conversion of Isometric Views to Orthographic Views	1	2
24	Floor plans of 2 or 3 rooms including windows, doors, and fixtures such as WC, bath, sink, shower, etc.	1	2

Suggested Text:

1. Bhatt N.D., Panchal V.M. & Ingle P.R., (2014), Engineering Drawing, Charotar Publishing House
2. Shah, M.B. & Rana B.C. (2008), Engineering Drawing and Computer Graphics, Pearson Education
3. S.N Lal, Engineering Drawing with Introduction to Auto CAD, Cengage Learning India Pvt Lid, New Delhi, 2018.
4. Agrawal B. & Agrawal C. M. (2012), Engineering Graphics, TMH Publication
5. Narayana, K.L. & P Kannaiah (2008), Text book on Engineering Drawing, Scitech Publishers
6. (Corresponding set of) CAD Software Theory and User Manuals

NOTE:

1. At least 20 sheets must be covered.
2. Sheet number 1 to 6 (Graph sheets / drawing sheets)
3. Sheet number 7 to 24 (AutoCAD drawings).